

Formulation and evaluation of Naproxen buccal films by using solvent casting method

Ms. N.S.Keerthanadevi^{*}, R.Bhuvaneshkumar, G.Bhuvaneshwari,
S.Chandirasekar

Submitted: 15-15-2023

Accepted: 25-12-2023

ABSTRACT:

The idea of mucoadhesion has attracted significant attention in the pharmaceuticals community in recent decades. Mucoadhesive buccal medication administration provides a number of benefits, including extending the dose's residence duration at the application site. The main aim of the study is to formulate and evaluate naproxen buccal films by using HPMCK15 as rate controlling polymer and poly vinyl pyrrolidone as an adhesive agent. Buccal films were prepared by using solvent casting method. The drug and drug-polymer compatibility was studied using FTIR studies. The results represent that the naproxen buccal films bypasses first-pass metabolism and reduce desirable side effects with desired adhesive properties. These films are proposed for the use of naproxen drug in buccal and dental inflammatory diseases.

INTRODUCTION

The situation in which two materials, at least one of which is biological in nature, are kept together for an extended amount of time by means of interfacial forces is known as bioadhesion or mucoadhesion. This idea started to be used with medicine delivery devices in the 1980s. Mucosal drug deliveries for drug transport are as follows:

- Buccal/oral
- Nasal
- Ocular
- Vaginal
- Rectal

Buccal delivery is the administration of medications through the mucosal membrane lining the cheeks. Buccal mucosa is suitable for delivering less permeable molecules and peptide medicines and considered to be the absolute route of mucoadhesive drug delivery system.

Naproxen is a non-steroidal anti-inflammatory drug meant for reducing ongoing inflammation, pain and fever. The aim of the study is to develop a buccal film containing a small dose

of naproxen meant for local delivery that can be easily applied and removed by the patient itself for the treatment of buccal inflammatory diseases such as periodontitis, stomatitis.

Mechanism Of Mucoadhesion

- ✓ Contact stage
- ✓ Consolidation stage

Contact stage: During this stage an intimate wetting, spreading and swelling of the mucoadhesive material creates a close contact between a bio adhesive and a mucosal membrane.

Consolidation stage: Moisture breaks molecules and causes attractive interaction between the two surfaces.

II. MATERIALS AND METHODS

HPMCK₁₅ was obtained as gift sample from colorcon asia pvt.ltd. Goa, Poly vinyl pyrrolidone was brought from Bangalore fine chemicals, Poly ethylene glycol was brought from Bangalore fine chemicals, Ethylcellulose, Glycerol, Propylene glycol were brought from isochem laboratories.

FORMULATION OF NAPROXEN FILMS BY SOLVENT CASTING METHOD

Solvent casting method is the most widely used method for the manufacture of buccal films.

The steps involved are as follows

- Hydrophilic polymers are dissolved in water to form a homogenous viscous solution.
- API and other lipophilic excipients are dissolved in suitable solvent to form a clear viscous solution.
- Both the solutions are mixed to form a clear homogenous mixture.
- The plate is lubricated evenly with glycerin and the mixture is poured in it.
- Then the resulting solution is allowed to dry and once it is dried the films are cut into needed size.

Formulation	Ingredients									
	Naproxen (mg)	HPMC K ₁₅ (mg)	PVP (mg)	PEG (mg)	Ethyl cellulose (mg)	Glycerol (ml)	Propylene glycol (ml)	Ethanol	Menthol	Sucrose
F ₁	100	200	100	20	20	1	-	q.s	q.s	q.s
F ₂	100	250	100	20	20	1	-			
F ₃	100	200	100	20	20	-	1			
F ₄	100	250	100	20	20	-	1			

Table 01: Formulation flow chart

EVALUATION OF NAPROXEN BUCCAL FILMS:

1. Weight of the film

Buccal film is weighed by calibrated weighing balance. Individual weight of each film is calculated. Average weight is calculated and analyzed.

2. Thickness

Thickness of buccal film is evaluated by calibrated micrometer screw gauge. The thickness is measured at five different points of the film and means value is calculated. This is done to ensure the uniformity in the thickness of the film as it is directly correlated with accuracy of dose in the film and supports the reproducibility of the method used for the formulation.

3. Film clarity

Examine the buccal film visually for any defects such as cracks, discoloration, or uneven distribution of the film material.

4. Light transmission test

Hold the buccal film against a light source to assess its transparency. Clarity is often

associated with the film's ability to transmit light uniformly

5. Surface smoothness

Run your fingers over the surface of the buccal film to check for smoothness. A rough or uneven surface may indicate a lack of clarity.

6. Folding endurance

The test is performed by folding the films numerous times repeatedly until the films begin to crack or break. The test is occasionally limited to a maximum of 300 folds, and the value is stated as the number of folds the film can withstand before rupturing.

7. FTIR spectral analysis

FTIR study was carried out to check the identity of drug and to determine the drug and excipient compatibility studies. Infrared spectrum of Naproxen was determined using Bruker's spectrophotometer. It involves passing light through a sample and measuring the intensity of the transmitted light at different wavelengths. This information is used to create a spectrum, which reveals details about the sample's composition.

III. RESULTS AND DISCUSSION

Weight of the film

The films are weighed individually and the average is taken.

Formulation	Individual weight (mg)	Average (mg)
F ₁	102.3	102.8
F ₂	101.8	
F ₃	105.9	
F ₄	101.4	

Table 02: Weight measurement of films

Thickness

The thicknesses of the films are calculated individually and the mean are taken into consideration.

Formulation	Individual thickness	Average
F ₁	0.1	0.1
F ₂	0.1	
F ₃	0.1	
F ₄	0.1	

Table 03: Thickness measurement of films

Film clarity

The films are evaluated under visual inspection and the results are declared.(Table – 4)

Surface smoothness

The films are evaluated for their surface smoothness and the results are declared. (Table – 4)

Light transmission test

The films are placed in front of light and the transmitted light is observed and the results are declared. (Table – 4)

Folding endurance

The films are folded for numerous times repeatedly and the time in which crack or braking of films occurs is recorded.

(Table – 4).

Formulation	Film clarity	Light transmission test	Surface smoothness	Folding endurance
F ₁	Clear	Clear	Smooth	250<
F ₂	Clear	Clear	Smooth	250<
F ₃	Clear	Clear	Smooth	250<
F ₄	Clear	Clear	Smooth	250<

Table 04: Physical properties of films

ANALYTICAL METHODS:

Determination of λ_{max} by using methanol

The maximum absorption for naproxen standard solution (250 $\mu\text{g/ml}$) was measured

between 200 – 400 nm by using methanol was found to be 230 nm and it is shown in the following figure

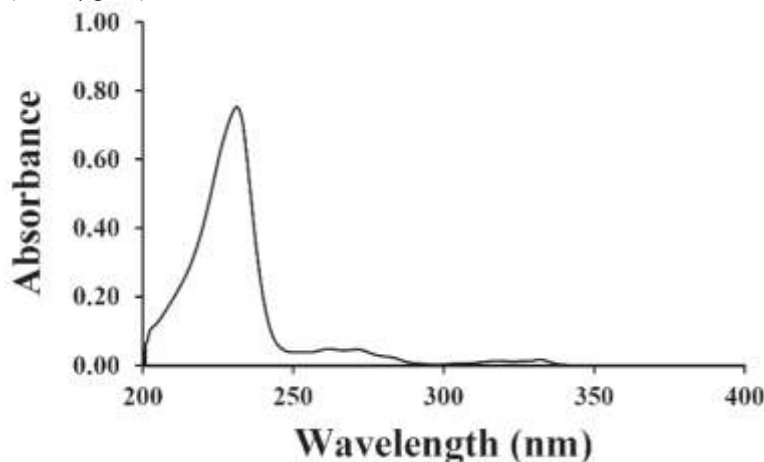


Figure 01: Absorption spectrum of Naproxen standard solution against Methanol blank.

Preparation of standard curve of naproxen in Methanol

UV absorption spectrum of naproxen in methanol showed λ_{max} at 230 nm. Absorbance obtained for various concentration of naproxen in

methanol are given in table. The graph of absorbance vs concentration for naproxen was found to be linear in the concentration range of 10 - 40 $\mu\text{g/ml}$. The drug obeys Beer – Lambert’s law in the range of 10 - 40 $\mu\text{g/ml}$.

S.No	Concentration ($\mu\text{g/ml}$)	Absorbance (nm)
1	10	1.15
2	20	1.96
3	30	2.81
4	40	3.68

Table 05: Data of absorbance vs concentration for naproxen

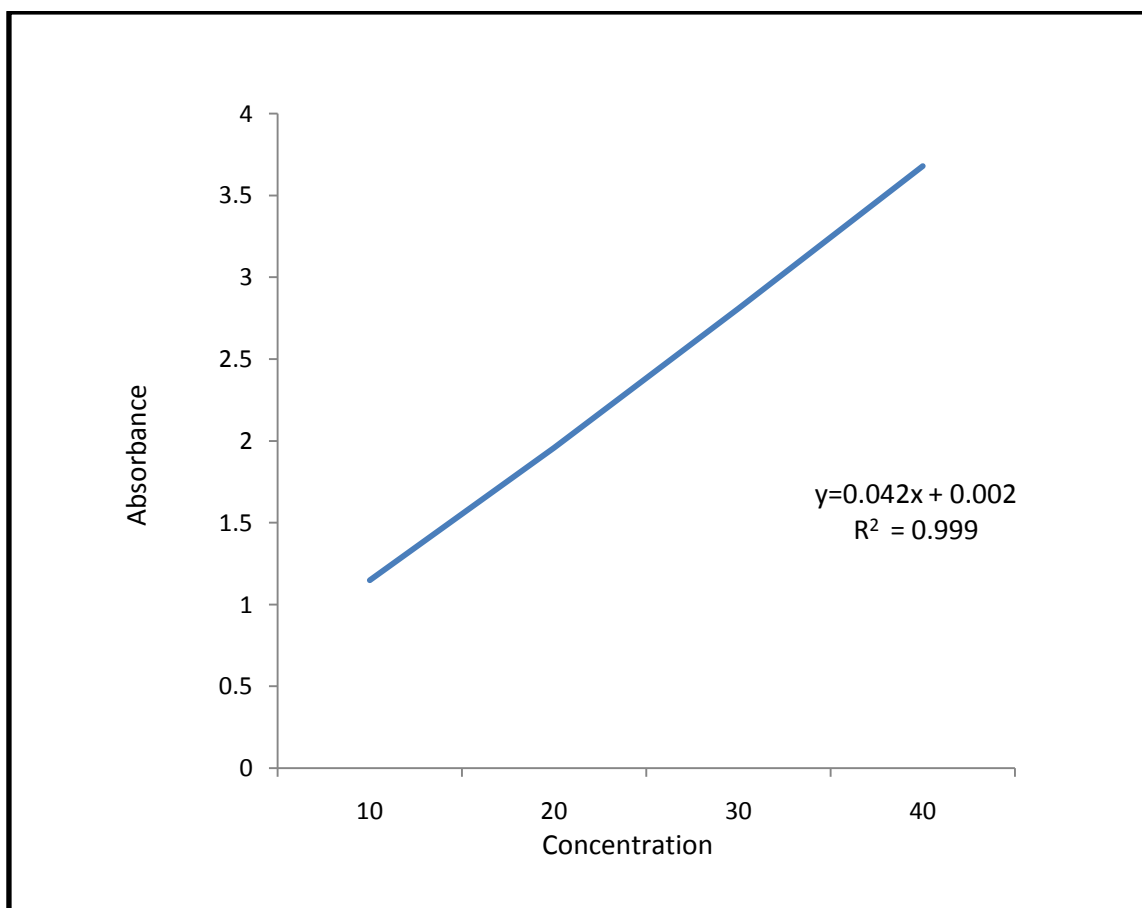


Figure 2: Standard graph of Naproxen in methanol

**FTIR Spectral Analysis:
Identification Of Naproxen By FTIR Spectroscopy:**

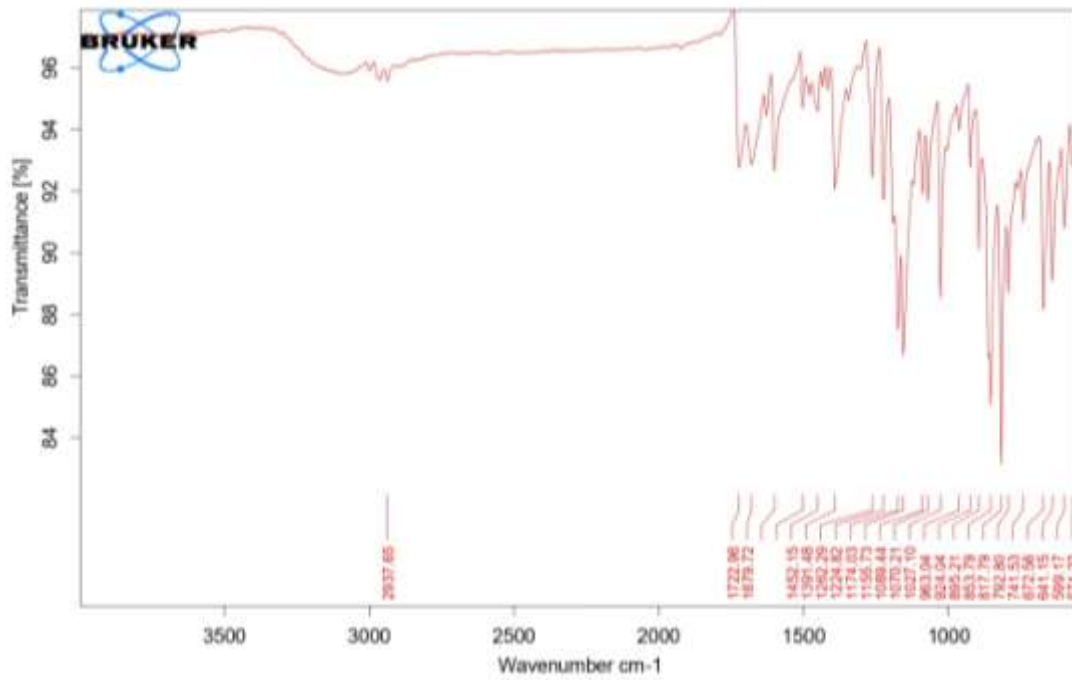


Figure: FTIR spectrum of pure drug Naproxen

Drug Excipient Compatibility Studies

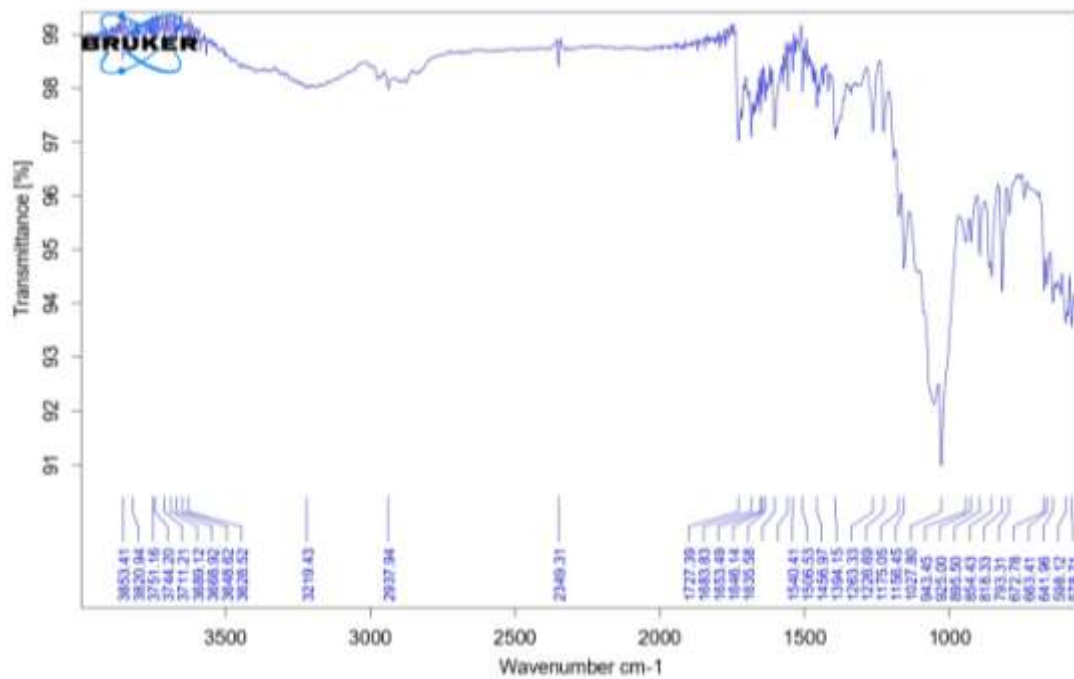


Figure FTIR spectrum of pure drug Naproxen and HPMC K15

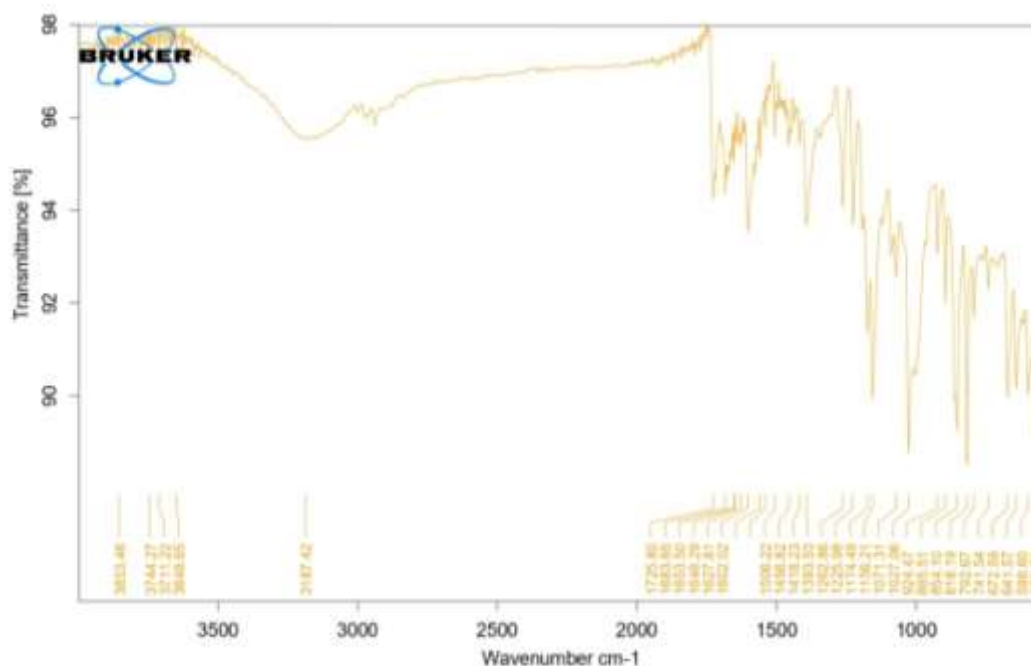


Figure : FTIR spectrum of pure drug Naproxen and ethyl cellulose

IV.CONCLUSION

The developed Naproxen mucoadhesive film containing HPMC polymer is satisfied in terms of good bioadhesive characteristics and controlled drug release thereby reducing the dosing frequency and prevents hepatic metabolism. In comparison of the formulations F₂ and F₄ shows extended drug release time.

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